

Prepared and presented by: Mr. Mohamad Seif





OBJECTIVES

1 Definition of electrostatic

2 Structure of atom

Charge of a body

Definition of electrostatic



Electrostatics: is a branch of electricity concerned with static (stationary) charges which accumulate on the surface of the materials

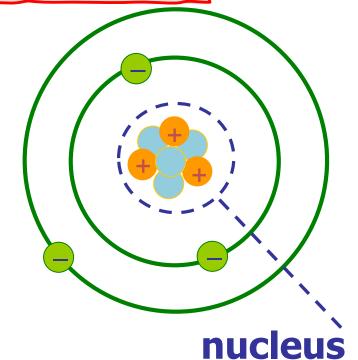






Matter is composed of atoms

- Each atom consists of a nucleus and electron cloud.
- The nucleus consists of two types of particles: protons which have a positive charge and neutrons which are neutral.
- The electrons are negatively charged and spread in the electron cloud around the nucleus.



Charge of a body



- The charge of an electron is $-e=-1.6\times 10^{-19}C$ and the charge of a proton is $e=1.6\times 10^{-19}C$.
- In neutral bodies, the number of electrons in each atom is equal to the number of protons.
- If an object loses electrons, it has a deficit of electrons. So, it becomes positively charged object.
- If an object gains electrons, it has an excess number of electrons. So, it becomes negatively charged object.

Charge of a body



Total Charge (Q) of a body:

If an object has an excess or deficit of N electrons, its total charge is given by:

 $Q = N \times e$ Be Smart

Where e is charge of one electron

Charge of a body



Application 1:

Consider a body containing 5.2×10^5 electrons. Calculate the total charge of the body.

$$|Q| = N \times e$$

$$|Q| = (5.2 \times 10^{5}) \times (1.6 \times 10^{-19}C)$$
 $& Smart$
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 $|Q| = (5.2 \times 10^{5}) \times (1.6 \times 10^{-19}C)$

Law of electric charges

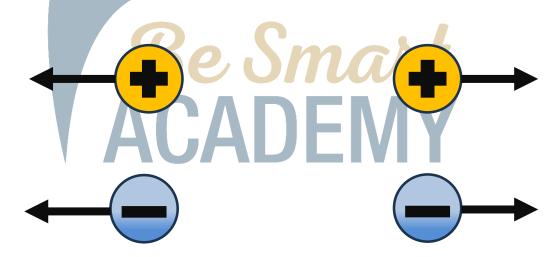


Unlike charges attract each other.





Like charges repel each other







OBJECTIVES

1 Types of Electrifications

2 Electrification of friction

3 Electrification by contact



Electrification: is the process of charging objects by adding or removing electrons.

Electrification is of three types:

- 1. Electrification (Charging) by friction.
- 2. Electrification (Charging) by contact (conduction).
- 3. Electrification (Charging) by induction



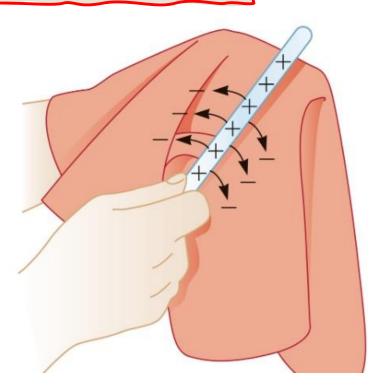


Electrification by friction:

If we rub an object (A) with another object (B), then:

Electrons transfer from one of them to the other:

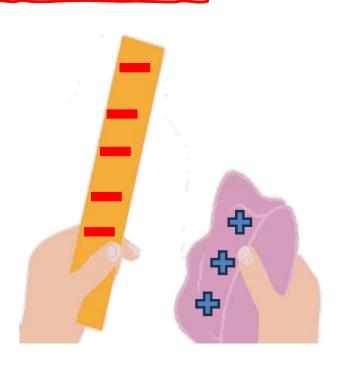
The two bodies to become charged with opposite charges. ACADE





If the electrons move from A to B, then:

Body (A) loses electrons (A has deficit of electrons) to become positively charged. Body (B) gains electrons (B has excess in electrons) to become negatively charged.



At the end of electrification by friction the two bodies (A) and (B) attract each other, since they are of unlike charges



Notes

During rubbing a piece of glass with a piece of silk, electrons transfer from the glass to silk

During rubbing a piece of plastic with a piece of fur, electrons transfer from the fur to plastic



Application 2:

An ebonite rod (Kind of rubber) is rubbed with a piece of fur.

Explain what happens?

After rubbing, the ebonite rod with fur, some electrons moves from the <u>fur</u> to the <u>ebonite rod</u>, then:

The ebonite rod acquires an excess of

electrons (negatively charged).

The fur acquires a deficit of electrons. (positively charged)

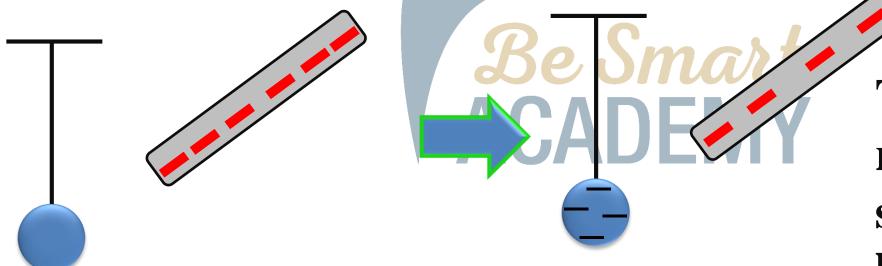


Electrification by contact (Conduction):

When a negatively charged rod comes in contact with a neutral ball:

Some electrons transfer from the rod to the neutral ball.

The two bodies become of same sign (negatively charged)



The two bodies repel each other since they are of like charges



Conduction



Electrons transfer to the object making it the same charge

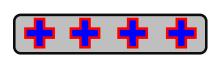


Application 3:

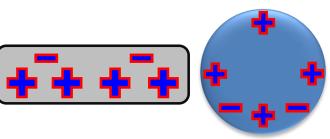
A positively charged glass rod in contact with the neutral ball Explain what happens?

After contact, some electrons move from the neutral ball to

the glass rod.









Neutral

Positively charged metal

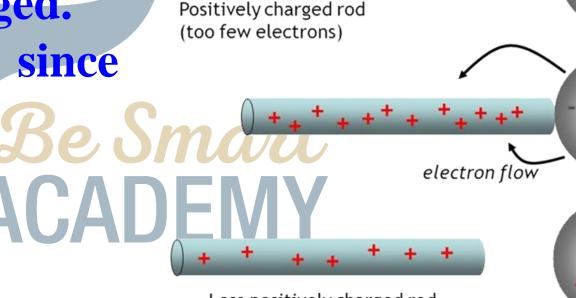
metal

For both the rod and the ball, the number of positive charges

is greater than negative charges

Both are positively charged.

They repel each other, since they have same charge



Less positively charged rod





OBJECTIVES

1 Electrification by Induction

Grounding

two objects.



Electrification by Induction:
No contact occurs between the Induction

The charged object influences a neutral object.

A Redistribution of charges (1) \(\)

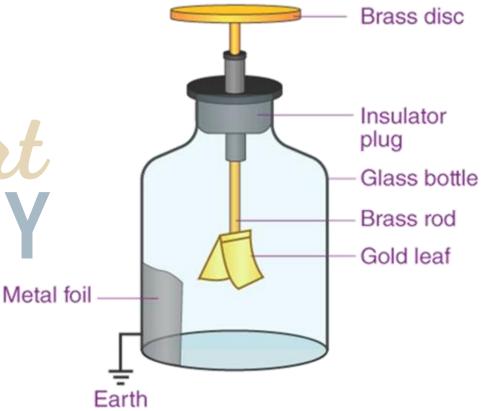
occur in the induced body (A) | Electrons Move Away Side Becomes Positive



Note:

An electroscope is used in different methods of electrifications.

The electroscope is an instrument used to detect the presence of electric charge on a body. CADEMY





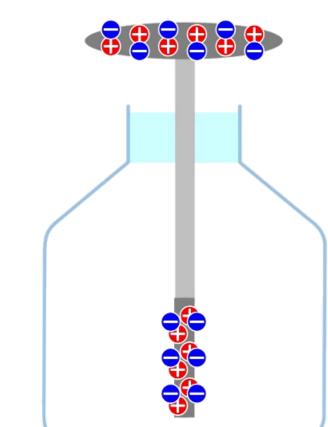
Application 4:

A negatively charged rod is set close to a neutral ball of the electroscope).

Explain what happens?

The electrons of the negatively charged rod repel the electrons on the neutral disc

Redistribution of charges occurs on the electroscope.





Some electrons move from the disc towards the leaves The disc become positively charged, while the leaves become negatively charged. The leaves then diverge. If the rod is moved away, the electroscope returns neutral by redistribution of charges.



Induction



Electrons move within the object making the closest side oppositely charged

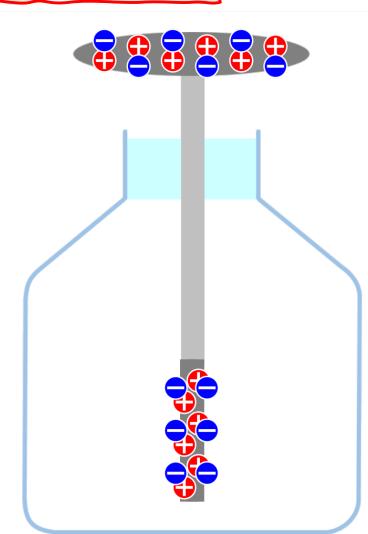


Application 5:

A positively charged rod is brought close to a neutral ball of the electroscope. Explain what happens?

The positive charges of the rod attract the electrons in the neutral disc & leaves.

The electrons move from the leaves to disc.



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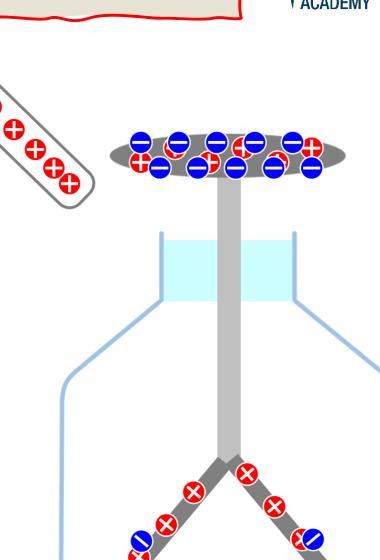




The negative charges arrange on the surface of the ball (Redistribution of charges).

The disc becomes negatively charged, while the leaves become positively charged

The leaves then diverge \(\) If the rod is moved away, the electroscope returns neutral by redistribution of charges



Grounding



A charged object is grounded when it is connected to the earth through a connecting wire.

When a body is grounded, it return neutral body.

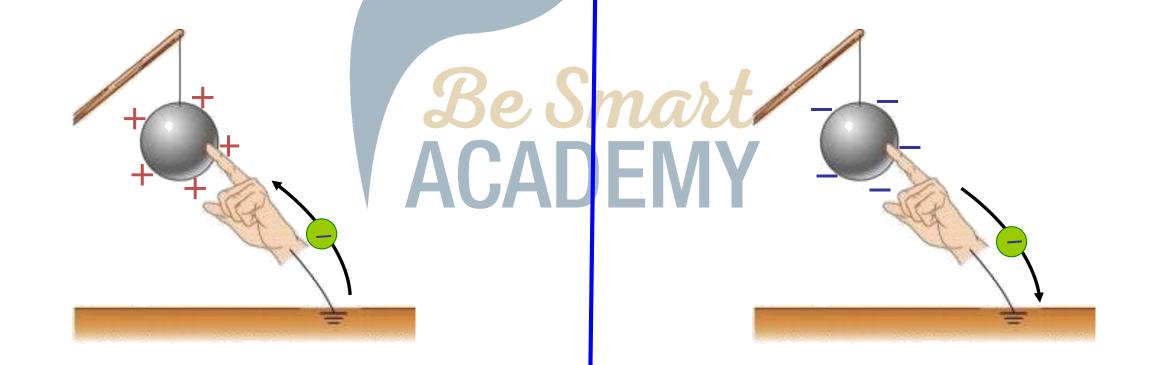
Grounding occurs by adding or removing electrons from the body.

Grounding



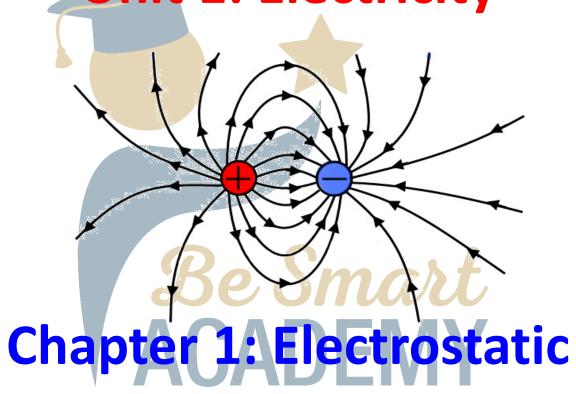
electrons flow from the <u>earth</u> to the metal ball to make it neutral.

electrons flow from the earth to the metal ball to make it neutral









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OBJECTIVES

1 Law of conservation of charges

Coulomb's law

Law of conservation of charges



Charge is never created or destroyed it is only transferred from one object to another.

The total number of charges of a body is conserved.



Law of conservation of charges



Application 6:

- Two identical charges A and B of charges $q_1 = -8 \times 10^{-6} C$ and $q_2 = 5 \times 10^{-6} C$ respectively.
- The two charges are placed in contact then separated after a short time.
- Calculate the charges of A and B after contact using law of conservation of charges A CAD E WY



Law of conservation of charges



$$q_1 = -8 \times 10^{-6}C; q_2 = 5 \times 10^{-6}C$$

Calculate the charges of A and B after contact the using law of conservation of charges

$$q_1 + q_2 = q_1' + q_2'$$

A B

But the two bodies are identical then: $q_1' \neq q_2'$

$$q_1 + q_2 = q'_1 + q'_1$$
 $q_1 + q_2 = q'_1 + q'_1$ $q_1 + q_2 = 2q'_1$

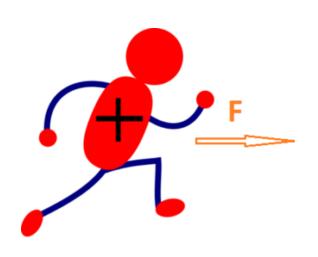
$$q_1' = q_2' = \frac{q_1 + q_2}{2} = \frac{-8 \times 10^{-6} + 5 \times 10^{-6}}{2} = -1.5 \times 10^{-6} C$$

Coulomb's law



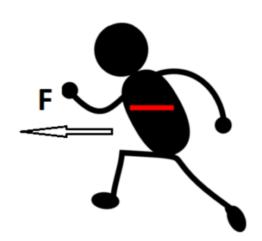
Statement of Coulomb's law:

The force of interaction between the two charges is proportional to the product of absolute values of the charges and inversely proportional to the square of the distance.



Be Smart ACADEMY

$$F = \frac{K|q_1|.|q_2|}{d^2}$$



Coulomb's law



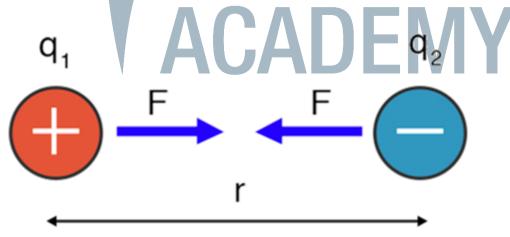
Application 7:

Given: Two-point charges (A) and B of charges $q_1 = 2 \times 10^{-5} C$ and $q_2 = -4 \times 10^{-5} C$ respectively.

The two-point charges are separated by 30 cm.

Given $K = 9 \times 10^9 S.I.$

Calculate the attractive force between the two charges.



Coulomb's law

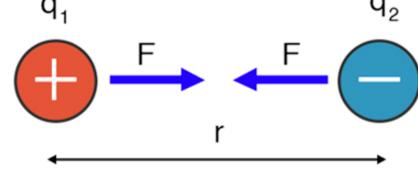


$$q_1 = 2 \times 10^{-5}C$$
; $q_2 = -4 \times 10^{-5}C$; $d = 0.3m$; $K = 9 \times 10^9 S.I$

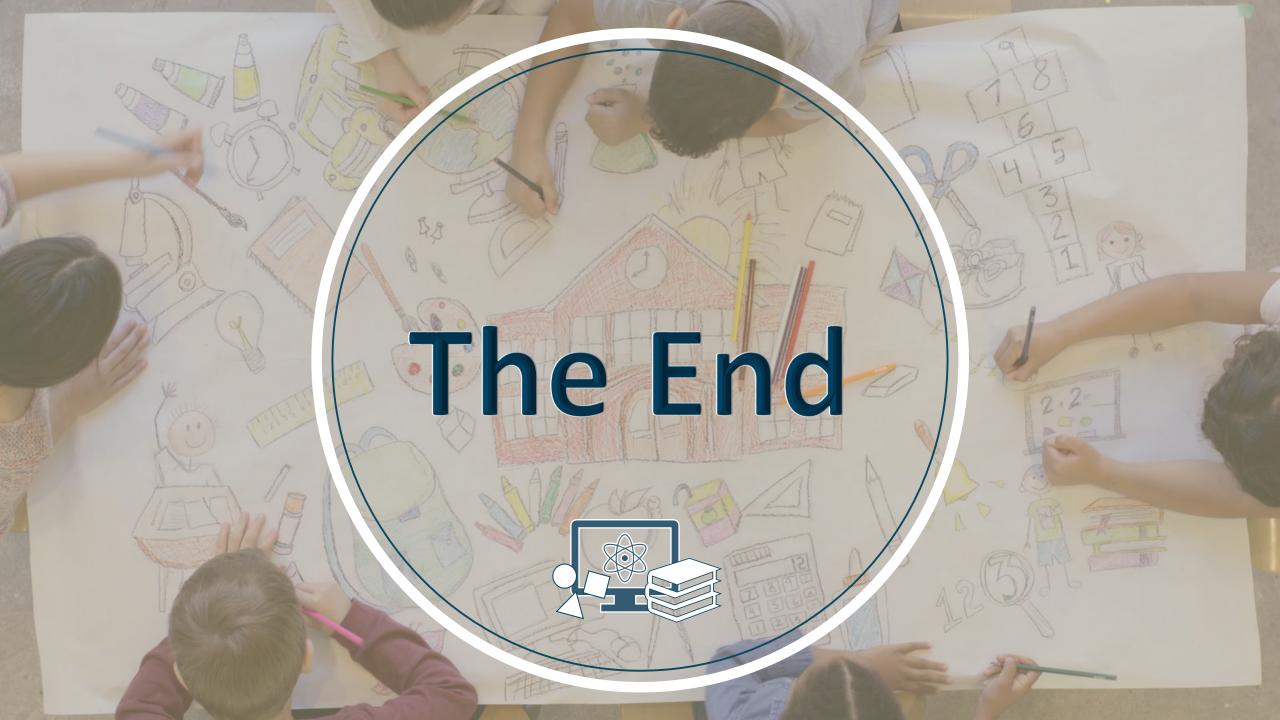
Calculate the attractive force between the two charges.

$$F = \frac{K|q_1|.|q_2|}{d^2}$$

$$F = \frac{9 \times 10^9|2 \times 10^{-5}|.|24 \times 10^{-5}C|^4}{(0.3)^2}$$



$$F = 80N$$



Be Smart



